





Characterization of Dy₂S₃ thin films deposited by successive ionic layer adsorption and reaction (SILAR) method

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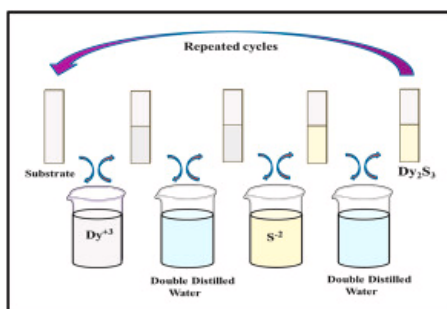
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Abstract

The successive ionic layer adsorption and reaction (SILAR) method was used for the deposition of dysprosium sulfide (Dy₂S₃) thin films on stainless steel (SS) substrate. The X-ray diffraction (XRD) study showed orthorhombic crystal structure of Dy₂S₃. Scanning electron microscopic study revealed microstructure with randomly distributed spherical nanostructured particles. The films exhibited hydrophilic nature with a contact angle of 50° and a specific surface area of 48 m²g⁻¹. The electrochemical properties of Dy₂S₃ films in 1 M Na₂SO₄ electrolyte displayed maximum specific capacitance (C_s) of 273 Fg⁻¹ at a scan rate of 5 mVs⁻¹.

Graphical abstract



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Introduction

Increasing energy demands contrasted with diminishing non-renewable energy sources invites a considerable intensification of efforts to develop environment-friendly energy storage devices. The growing need for batteries, supercapacitors, and fuel cells becomes more pronounced [1,2]. Energy storage systems consisting of batteries have low life cycles and need larger space for system [3]. On the other hand, supercapacitors (SCs) with remarkable