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## Enhanced energy density of flexible asymmetric solid state supercapacitor device fabricated with amorphous thin film electrode materials

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## **Abstract**

Supercapacitors have recently received immense interest in scientific community, as a complementary technology to batteries, to meet the various requirements for energy usage in practice. Amorphous MnO<sub>2</sub> and CuS thin films are prepared on stainless steel-304 (SS) substrate by chemical bath deposition (CBD) and successive ionic layer adsorption and reaction (SILAR) methods, respectively. Asymmetric flexible solid state supercapacitor fabricated with configuration of SS/A-MnO<sub>2</sub>/Polyvinyl acetate (PVA)-Na<sub>2</sub>SO<sub>4</sub>/A-CuS/SS delivered an impressive specific energy of 57.4Wh kg<sup>-1</sup> at specific power 317Wkg<sup>-1</sup> and excellent cycling stability over 10,000 cycles with capacitive retention of 84%. Moreover, series configuration of two asymmetric devices shows the capability of powering 211 red LEDs for ~150s after charging for 30s.

## Graphical abstract

(A) FE-SEM of amorphous MnO<sub>2</sub>, (B) The XRD graphs of <u>Amorphous</u> MnO<sub>2</sub> and CuS, (C) Comparision of energy density and power density of SS/A-CuS/Polyvinyl acetate (PVA)-Na<sub>2</sub>SO<sub>4</sub>/A-MnO<sub>2</sub>/SS (D) Assembly of 5×5cm<sup>2</sup> SS/A-CuS/Polyvinyl acetate (PVA)-Na<sub>2</sub>SO<sub>4</sub>/A-MnO<sub>2</sub>/SS solid state flexible asymmetric supercapacitor device with amorphous MnO<sub>2</sub> as an anode and amorphous CuS as a cathode, and (E) FE-SEM of amorphous CuS, and (F) Stability graph of supercapacitor device.